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cis-Verbenol

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cis-Verbenol (alternative name: 4,6,6-trimethylbicyclo[3.1.1]-hept-3-en-2-ol), $C_{10}H_{16}O$, forms an orthorhombic $P2_12_12_1$ crystal that contains three molecules per asymmetric unit. These three molecules form hydrogen-bonded helices parallel to the shortest axis of the lattice. The $O \cdots O$ distances associated with the hydrogen bonds are 2.760 (3), 2.760 (3) and 2.766 (3) Å.

Comment

Verbenol is a monoterpene alcohol with four known stereoisomers (*Dictionary of Natural Products*, 1994). Two enantiomers, (1R,2S,5R) and (1S,2R,5S), are oils and are diastereoisomers of the solid enantiomeric structures, (1R,2R,5R) and (1S,2S,5S), studied here. The numbering and R/S notation used in the text refers to the 'chemical' numbering scheme. Recent solid-state NMR studies have demonstrated that verbenol exhibits multiple resonances per carbon position in the molecular structure, suggesting polymorphism or multiple molecules per asymmetric unit (Harper & Grant, 2000). An X-ray analysis was performed on (I) to clarify these structural variations and to characterize the associated hydrogen-bonding features.



The arrangement of the three molecules of the asymmetric unit for (I) is shown in Fig. 1. The crystal of (I) consists of helical hydrogen-bonded chains lying parallel to the shortest axis. The three distinct molecules per asymmetric unit differ primarily in the conformations of the hydrogen-bonded region of verbenol, with C3–C4–O–H dihedral angles of 55, 69 and 74°. All other differences in heavy-atom angles, both dihedral and geminal, and in bond lengths were modest between the three unique verbenol molecules, with respective variations of less than 2° and 0.013 Å. Four asymmetric units compose the unit-cell arrangement shown in Fig. 2.

Monofunctional alcohols frequently form chains, rings and helices. Such arrangements often contain more than one molecule per asymmetric unit (Brock & Duncan, 1994). The



Figure 1

The arrangement of the three *cis*-verbenol molecules in the asymmetric unit of (I) showing the relative configuration. Displacement ellipsoids are drawn at the 50% probability level and H atoms are shown as spheres of arbitrary radii. The 'crystallographic' numbering differs from the 'chemical' numbering and the chemical (1S,2S,5S/1R,2R,5R) notation would be (1R,4R,5R/1S,4S,5S) in the crystallographic numbering scheme shown here.

present X-ray structure of verbenol corresponds to a previously described sterol monoalcohol structure (Brock *et al.*, 1994) in the number of molecules per asymmetric unit, general hydrogen-bonding arrangement and space group. Similar hydrogen-bonded helices containing three molecules per asymmetric unit have also been observed in other monoalcohols (Singelenberg & van Eijck, 1987; Zavodnik *et*



Figure 2

The arrangement of verbenol in the unit cell of (I) viewed along the a axis. Extension of the helical structure illustrates the propagation of the coil beyond a given asymmetric unit.

al., 1987; Escobar & Wittke, 1984), although with different space groups. Solid-state NMR analyses demonstrate that other preparations of solid verbenol powder samples contain a second minor polymorph with four molecules per asymmetric unit (Harper & Grant, 2000). A single crystal of this second structure could not be isolated and hence has not been subjected to X-ray analysis.

Experimental

Verbenol was obtained from Aldrich as a 50+% enantiomeric excess (1S,2S,5S/1R,2R,5R) mixture and used as received. Reported enantiomeric excess was determined by Aldrich using chiral stationary phase gas chromatography. A 0.6/0.4 (s.u. 0.1) (1S,2S,5S/1R,2R,5R) enantiomeric ratio was determined in our lab by polarimetry on a sample dissolved in acetone. The enantiomeric content is a measure of the bulk material and a corresponding value for the individual crystal could not be determined with the methods used. A verbenol purity of 94% was found by gas chromatography on an achiral stationary phase. The enantiomeric (1S,2S,5S/1R,2R,5R) mixture of (I) spontaneously resolved during crystallization by slow evaporation of a solution in 100% methanol and suitable crystals of one of the enantiomers of (I) (m.p. 341.5–346.5 K) were obtained.

Crystal data

| $C_{10}H_{16}O$ $M_r = 152.23$ Orthorhombic, $P2_12_12_1$ $a = 7.0115 (1) Å$ $b = 18.7691 (6) Å$ $c = 21.1681 (7) Å$ $V = 2785.71 (13) Å^3$ $Z = 12$ $D_x = 1.089 \text{ Mg m}^{-3}$ | Mo $K\alpha$ radiation Cell parameters from 252 reflections $\theta = 4.10-24.97^{\circ}$ $\mu = 0.068 \text{ mm}^{-1}$ T = 200 (1) K Prism, colorless $0.20 \times 0.14 \times 0.13 \text{ mm}$ |
|--|--|
| Data collection | |
| Nonius KappaCCD diffractometer φ and ω scans 13 724 measured reflections 2717 independent reflections 2360 reflections with $I > 2\sigma(I)$ $R_{int} = 0.042$ | $\theta_{max} = 24.79^{\circ}$ $h = -8 \rightarrow 7$ $k = -22 \rightarrow 22$ $l = -24 \rightarrow 24$ Intensity decay: <2% |
| Refinement | |
| Refinement on F^2 R(F) = 0.047 $wR(F^2) = 0.127$ S = 1.05 2717 reflections 310 parameters H atoms treated by a mixture of independent and constrained refinement | $\begin{split} &w = 1/[\sigma^2(F_o{}^2) + (0.0812P)^2 \\ &+ 0.6806P] \\ &where \ P = (F_o{}^2 + 2F_c{}^2)/3 \\ (\Delta/\sigma)_{\max} < 0.001 \\ \Delta\rho_{\max} = 0.24 \text{ e} \text{ Å}{}^{-3} \\ \Delta\rho_{\min} = -0.23 \text{ e} \text{ Å}{}^{-3} \end{split}$ |
| Table 1 | |
| Selected geometric parameters (Å, $^{\circ}$). | |

| 01-C4 01'-C4' | 1.444 (3) 1.443 (4) | O1″-C4″ | 1.443 (4) |
|-------------------------|------------------------|-------------|-----------|
| O1-C4-C3 O1'-C4'-C3' | 111.5 (3) 111.7 (3) | O1″-C4″-C3″ | 111.0 (3) |

Table 2

Hydrogen-bonding geometry (Å, $^{\circ}$).

| $D-\mathrm{H}\cdots A$ | D-H | $H \cdot \cdot \cdot A$ | $D \cdots A$ | $D - H \cdots A$ |
|------------------------------|----------|-------------------------|--------------|------------------|
| $01 - HO1 \cdots O1'$ | 0.81 (4) | 1.97 (4) | 2.760 (3) | 162 (4) |
| $01' - HO1' \cdots O1''$ | 0.70 (4) | 2.11 (5) | 2.760 (3) | 156 (5) |
| $01'' - HO1'' \cdots O1^{i}$ | 0.75 (5) | 2.02 (5) | 2.766 (3) | 170 (4) |

Symmetry code: (i) x - 1, y, z.

Hydroxy H atoms were located and refined isotropically. All other H atoms were refined as riding models on the appropriate C atoms using *SHELXL*97 (Sheldrick, 1997) restraints and assigned isotropic displacement parameters of 1.5 for methyl and 1.2 for all other H atoms. The lack of suitable anomalous scatterers did not allow us to determine the absolute configuration of the enantiomer studied and we report here the relative configuration of *cis*-verbenol.

Data collection: *COLLECT* (Nonius, 1998); cell refinement: *DENZO-SMN* (Otwinowski & Minor, 1997); data reduction: *DENZO-SMN*; program(s) used to solve structure: *SIR*97 (Altomare *et al.*, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *WinGX* (Farrugia, 1998) and *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *ORTEP-3 for Windows*.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: BK1491). Services for accessing these data are described at the back of the journal.

References

- Altomare, A., Cascarano, C., Giacovazzo, C., Guagliardi, A., Moliterni, A. G. G., Burla, M. C., Polidori, G., Camalli, M. & Spagna, R. (1997). SIR97. University of Bari, Italy.
- Brock, C. P. & Duncan, L. L. (1994). Chem. Mater. 6, 1307–1312.
- Brock, C. P., Stoilov, I. & Watt, D. S. (1994). Acta Cryst. C50, 434-438.
- Dictionary of Natural Products (1994). Edited by J. Buckingham, Vol. 4, p. 4686. London: Chapman and Hall.
- Escobar, C. & Wittke, O. (1984). Acta Cryst. C40, 1469-1471.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Farrugia, L. J. (1998). WinGX. University of Glasgow, Scotland.
- Harper, J. K. & Grant, D. M. (2000). J. Am. Chem. Soc. In the press.
- Nonius (1998). COLLECT. Nonius BV, Delft, The Netherlands.
- Otwinowski, Z. & Minor, W. (1997). Methods Enzymol. 276, 307-326.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
- Singelenberg, F. A. J. & van Eijck, B. P. (1987). Acta Cryst. C43, 693-695.
- Zavodnik, V. E., Bel'skii, V. K. & Zorkii, P. M. (1987). Zh. Strukt. Khim. 28, 175–177.