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## Crystal Structure

## Communications

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## cis-Verbenol

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cis-Verbenol (alternative name: 4,6,6-trimethylbicyclo[3.1.1]-hept-3-en-2-ol), $\mathrm{C}_{10} \mathrm{H}_{16} \mathrm{O}$, forms an orthorhombic $P 2_{1} 2_{1} 2_{1}$ crystal that contains three molecules per asymmetric unit. These three molecules form hydrogen-bonded helices parallel to the shortest axis of the lattice. The $\mathrm{O} \cdots \mathrm{O}$ distances associated with the hydrogen bonds are 2.760 (3), 2.760 (3) and 2.766 (3) $\AA$.

## Comment

Verbenol is a monoterpene alcohol with four known stereoisomers (Dictionary of Natural Products, 1994). Two enantiomers, $(1 R, 2 S, 5 R)$ and $(1 S, 2 R, 5 S)$, are oils and are diastereoisomers of the solid enantiomeric structures, $(1 R, 2 R, 5 R)$ and $(1 S, 2 S, 5 S)$, studied here. The numbering and $R / S$ notation used in the text refers to the 'chemical' numbering scheme. Recent solid-state NMR studies have demonstrated that verbenol exhibits multiple resonances per carbon position in the molecular structure, suggesting polymorphism or multiple molecules per asymmetric unit (Harper \& Grant, 2000). An X-ray analysis was performed on (I) to clarify these structural variations and to characterize the associated hydrogen-bonding features.

(I)

The arrangement of the three molecules of the asymmetric unit for (I) is shown in Fig. 1. The crystal of (I) consists of helical hydrogen-bonded chains lying parallel to the shortest axis. The three distinct molecules per asymmetric unit differ primarily in the conformations of the hydrogen-bonded region of verbenol, with $\mathrm{C} 3-\mathrm{C} 4-\mathrm{O}-\mathrm{H}$ dihedral angles of 55,69 and $74^{\circ}$. All other differences in heavy-atom angles, both dihedral and geminal, and in bond lengths were modest between the three unique verbenol molecules, with respective variations of less than $2^{\circ}$ and $0.013 \AA$. Four asymmetric units compose the unit-cell arrangement shown in Fig. 2.

Monofunctional alcohols frequently form chains, rings and helices. Such arrangements often contain more than one molecule per asymmetric unit (Brock \& Duncan, 1994). The


Figure 1
The arrangement of the three cis-verbenol molecules in the asymmetric unit of (I) showing the relative configuration. Displacement ellipsoids are drawn at the $50 \%$ probability level and H atoms are shown as spheres of arbitrary radii. The 'crystallographic' numbering differs from the 'chemical' numbering and the chemical ( $1 S, 2 S, 5 S / 1 R, 2 R, 5 R$ ) notation would be $(1 R, 4 R, 5 R / 1 S, 4 S, 5 S)$ in the crystallographic numbering scheme shown here.
present X-ray structure of verbenol corresponds to a previously described sterol monoalcohol structure (Brock et al., 1994) in the number of molecules per asymmetric unit, general hydrogen-bonding arrangement and space group. Similar hydrogen-bonded helices containing three molecules per asymmetric unit have also been observed in other monoalcohols (Singelenberg \& van Eijck, 1987; Zavodnik et


Figure 2
The arrangement of verbenol in the unit cell of (I) viewed along the $a$ axis. Extension of the helical structure illustrates the propagation of the coil beyond a given asymmetric unit.
al., 1987; Escobar \& Wittke, 1984), although with different space groups. Solid-state NMR analyses demonstrate that other preparations of solid verbenol powder samples contain a second minor polymorph with four molecules per asymmetric unit (Harper \& Grant, 2000). A single crystal of this second structure could not be isolated and hence has not been subjected to X-ray analysis.

## Experimental

Verbenol was obtained from Aldrich as a $50+\%$ enantiomeric excess $(1 S, 2 S, 5 S / 1 R, 2 R, 5 R)$ mixture and used as received. Reported enantiomeric excess was determined by Aldrich using chiral stationary phase gas chromatography. A $0.6 / 0.4$ (s.u. 0.1 ) $(1 S, 2 S, 5 S / 1 R, 2 R, 5 R)$ enantiomeric ratio was determined in our lab by polarimetry on a sample dissolved in acetone. The enantiomeric content is a measure of the bulk material and a corresponding value for the individual crystal could not be determined with the methods used. A verbenol purity of $94 \%$ was found by gas chromatography on an achiral stationary phase. The enantiomeric $(1 S, 2 S, 5 S / 1 R, 2 R, 5 R)$ mixture of (I) spontaneously resolved during crystallization by slow evaporation of a solution in $100 \%$ methanol and suitable crystals of one of the enantiomers of (I) (m.p. 341.5-346.5 K) were obtained.

## Crystal data

$\mathrm{C}_{10} \mathrm{H}_{16} \mathrm{O}$
$M_{r}=152.23$
Orthorhombic, $P 2_{1} 2_{1} 2_{1}$
$a=7.0115$ (1) $\AA$
$b=18.7691$ (6) $\AA$
$c=21.1681$ (7) $\AA$
$V=2785.71(13) \AA^{3}$
$Z=12$
$D_{x}=1.089 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
Cell parameters from 252 reflections
$\theta=4.10-24.97^{\circ}$
$\mu=0.068 \mathrm{~mm}^{-1}$
$T=200$ (1) K
Prism, colorless
$0.20 \times 0.14 \times 0.13 \mathrm{~mm}$

## Data collection

Nonius KappaCCD diffractometer $\varphi$ and $\omega$ scans
13724 measured reflections
2717 independent reflections
2360 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.042$
$\theta_{\text {max }}=24.79^{\circ}$
$h=-8 \rightarrow 7$
$k=-22 \rightarrow 22$
$l=-24 \rightarrow 24$
Intensity decay: <2\%

## Refinement

Refinement on $F^{2}$
$R(F)=0.047$
$w R\left(F^{2}\right)=0.127$
$S=1.05$
2717 reflections
310 parameters
H atoms treated by a mixture of independent and constrained refinement

$$
\begin{aligned}
& \begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{o}^{2}\right)+(0.0812 P)^{2}\right. \\
&+0.6806 P] \\
& \quad \text { where } P=\left(F_{o}^{2}+2 F_{c}^{2}\right) / 3 \\
&(\Delta / \sigma)_{\max }<0.001 \\
& \Delta \rho_{\max }=0.24 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.23 \mathrm{e} \AA^{-3}
\end{aligned}
\end{aligned}
$$

Table 1
Selected geometric parameters ( $\mathrm{A},{ }^{\circ}$ ).

| $\mathrm{O} 1-\mathrm{C} 4$ | $1.444(3)$ | $\mathrm{O} 1^{\prime \prime}-\mathrm{C} 4^{\prime \prime}$ | $1.443(4)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{O} 1^{\prime}-\mathrm{C} 4^{\prime}$ | $1.443(4)$ |  |  |
| $\mathrm{O} 1-\mathrm{C} 4-\mathrm{C} 3$ | $111.5(3)$ | $\mathrm{O} 1^{\prime \prime}-\mathrm{C}^{\prime \prime}-\mathrm{C}^{\prime \prime}$ | $111.0(3)$ |
| $\mathrm{O}^{\prime}-\mathrm{C} 4^{\prime}-\mathrm{C} 3^{\prime}$ | $111.7(3)$ |  |  |

Table 2
Hydrogen-bonding geometry ( $\left(\AA^{\circ}{ }^{\circ}\right.$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{O} 1-\mathrm{HO} 1 \cdots \mathrm{O}^{\prime}$ | 0.81 (4) | 1.97 (4) | 2.760 (3) | 162 (4) |
| $\mathrm{O} 1^{\prime}-\mathrm{HO}^{\prime} \cdots{ }^{\prime} \cdot \mathrm{O}^{\prime \prime}$ | 0.70 (4) | 2.11 (5) | 2.760 (3) | 156 (5) |
| $\mathrm{O} 1^{\prime \prime}-\mathrm{HO1}^{\prime \prime} \ldots \mathrm{Ol}^{\text {i }}$ | 0.75 (5) | 2.02 (5) | 2.766 (3) | 170 (4) |

Symmetry code: (i) $x-1, y, z$.

Hydroxy H atoms were located and refined isotropically. All other H atoms were refined as riding models on the appropriate C atoms using SHELXL97 (Sheldrick, 1997) restraints and assigned isotropic displacement parameters of 1.5 for methyl and 1.2 for all other H atoms. The lack of suitable anomalous scatterers did not allow us to determine the absolute configuration of the enantiomer studied and we report here the relative configuration of cis-verbenol.

Data collection: COLLECT (Nonius, 1998); cell refinement: DENZO-SMN (Otwinowski \& Minor, 1997); data reduction: DENZO-SMN; program(s) used to solve structure: SIR97 (Altomare et al., 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: WinGX (Farrugia, 1998) and ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: ORTEP-3 for Windows.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: BK1491). Services for accessing these data are described at the back of the journal.

## References

Altomare, A., Cascarano, C., Giacovazzo, C., Guagliardi, A., Moliterni, A. G. G., Burla, M. C., Polidori, G., Camalli, M. \& Spagna, R. (1997). SIR97. University of Bari, Italy.
Brock, C. P. \& Duncan, L. L. (1994). Chem. Mater. 6, 1307-1312.
Brock, C. P., Stoilov, I. \& Watt, D. S. (1994). Acta Cryst. C50, 434-438.
Dictionary of Natural Products (1994). Edited by J. Buckingham, Vol. 4, p. 4686. London: Chapman and Hall.

Escobar, C. \& Wittke, O. (1984). Acta Cryst. C40, 1469-1471.
Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
Farrugia, L. J. (1998). WinGX. University of Glasgow, Scotland.
Harper, J. K. \& Grant, D. M. (2000). J. Am. Chem. Soc. In the press. Nonius (1998). COLLECT. Nonius BV, Delft, The Netherlands. Otwinowski, Z. \& Minor, W. (1997). Methods Enzymol. 276, 307-326. Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany. Singelenberg, F. A. J. \& van Eijck, B. P. (1987). Acta Cryst. C43, 693-695.
Zavodnik, V. E., Bel'skii, V. K. \& Zorkii, P. M. (1987). Zh. Strukt. Khim. 28, 175-177.

